

A Current Hot Topic in Regulated Bioanalysis: *“Impact of Hemolysis on Drug Stability”*

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Introduction

- The effect of hemolysis on bioanalytical methods **accuracy & precision** has been already discussed in literature.
- In general, simple modifications to the bioanalytical method usually resolve the effect of hemolysis: **Extraction and/or Chromatography**
- However, when the effect of hemolysis impacts **drug stability**, the solution can be more complicated.
- **Processed Reconstituted Stability (PRS)** of morphine extracted from hemolyzed plasma

The Issue

- During morphine **matrix effect evaluation**, a lower peak area response (**15% lower**) was observed for morphine and its deuterated internal standard for the hemolysed plasma samples when compared to non-hemolysed plasma
- When re-injected after a **PRS** of **16 hours**, the difference in peak area response was more significant (**50% lower**) when comparing the extracted hemolysed sample to extracted plasma.
- However, the **peak area ratio** was accurate for both experiments (Matrix effect & PRS).

Matrix Effect Evaluation of Morphine and PRS

	First injection (Freshly extracted batch)				Re-Injection (16 hours at 4°C)			
	LOW QC (0.75 ng/mL) n=3		HIGH QC (45.00 ng/mL) n=3		LOW QC (0.75 ng/mL) n=3		HIGH QC (45.00 ng/mL) n=3	
	Peak area	Area ratio	Peak area	Area ratio	Peak area	Area ratio	Peak area	Area ratio
Human plasma								
Mean	14072	0.1468	815461	9.0027	28472	0.1521	1532539	8.9215
CV (%)	3.6	3.5	2.0	0.8	2.8	0.9	4.0	2.8
Human plasma containing 7.5% hemolyzed whole blood								
Mean	11996	0.1453	677628	8.9790	11521	0.1489	762886	8.8229
CV (%)	1.4	3.8	0.7	1.0	15.6	2.9	6.3	1.2
Difference (%)	14.8	1.0	16.9	0.3	59.5	2.1	50.2	1.1

CV: Coefficient of variation; QC: Quality control

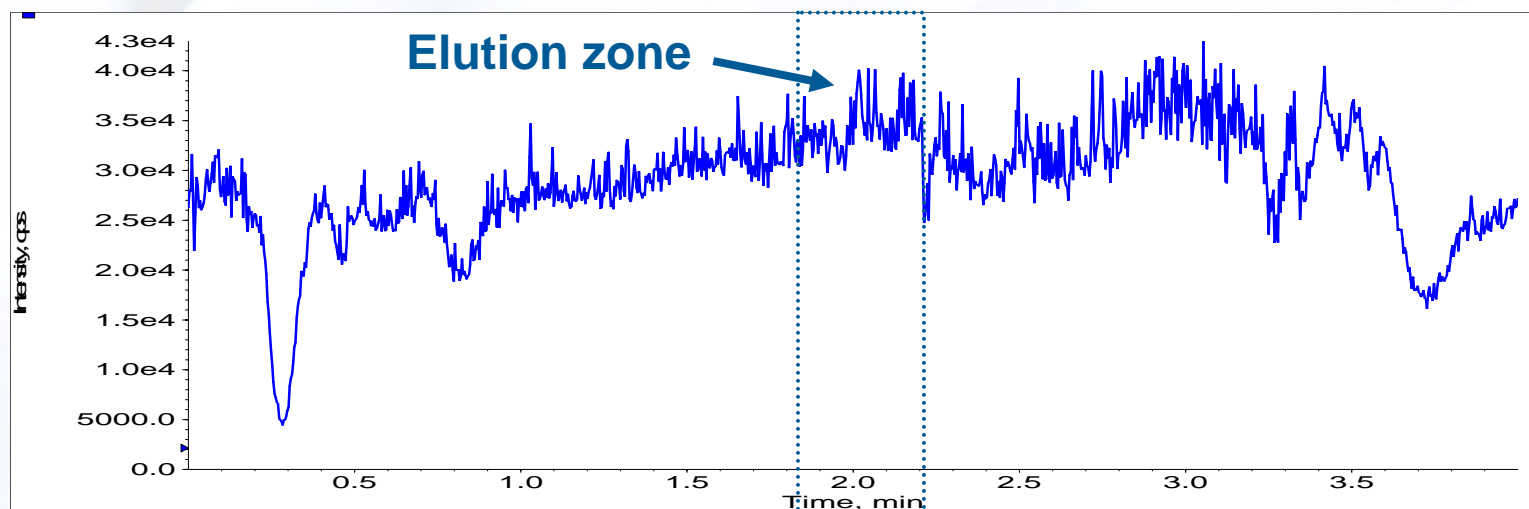
% Difference = ((plasma value - hemolyzed value) / plasma values) x 100

Matrix Effect Evaluation of Morphine

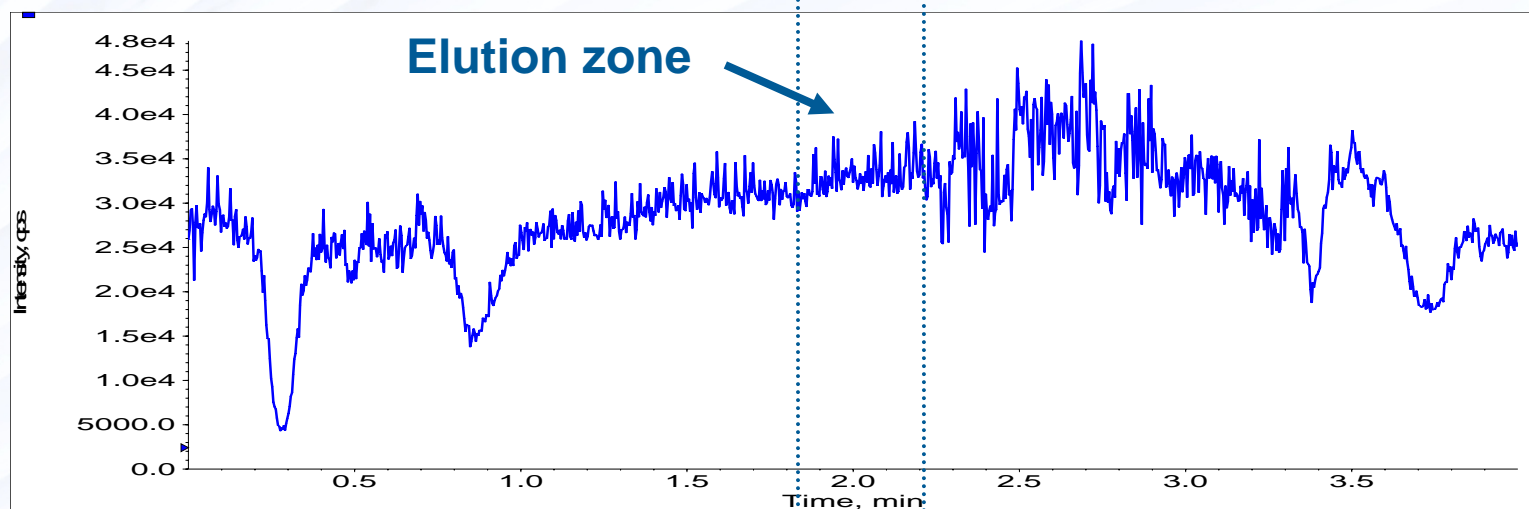
- *Is it a matrix effect / ion suppression?*
- Post-column infusion profile performed with freshly extracted hemolyzed blank samples revealed that **no ion suppression** cause the decrease of analyte and IS response.
- Possible appearance **over time** of a suppressor in the reconstitution solution was evaluated following **50 hours of storage**.
- The ionization profile of the extracted blank plasma and blank hemolyzed plasma sample were similar and showed no suppression.

Matrix Effect Evaluation of Morphine

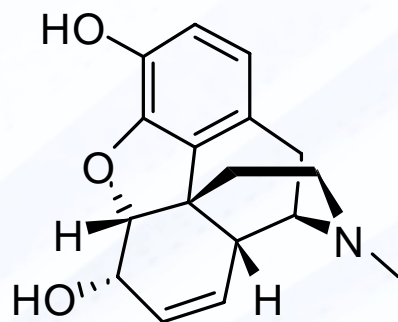
A Control plasma



B Hemolyzed plasma



PRS of morphine extracted from hemolyzed plasma



How would you have fixed this issue?

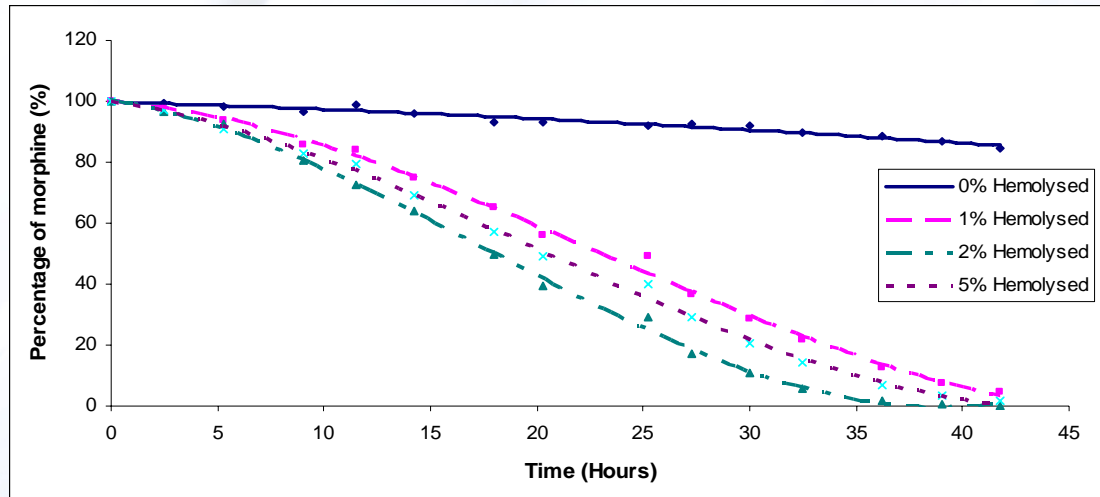
Initial Investigation

- Ionization suppression hypothesis was **eliminated**
- **PRS** issue of morphine occurring only in extracted hemolyzed samples was **considered**.
- Human plasma samples prepared at different levels of hemolysis **(0%,1%, 2% and 5%)**
- Since the reconstitution solution could promote degradation, two different solutions were evaluated.
 - MeOH / Water
 - MeOH / 20mM ammonium bicarbonate pH 10
- To confirm the **degradation trend** in the hemolyzed sample, extracted blank and hemolyzed blank plasma samples were reconstituted with a **pure solution** of morphine prepared in both reconstitution solutions and injected over time.

Morphine PRS at Different Hemolysis %

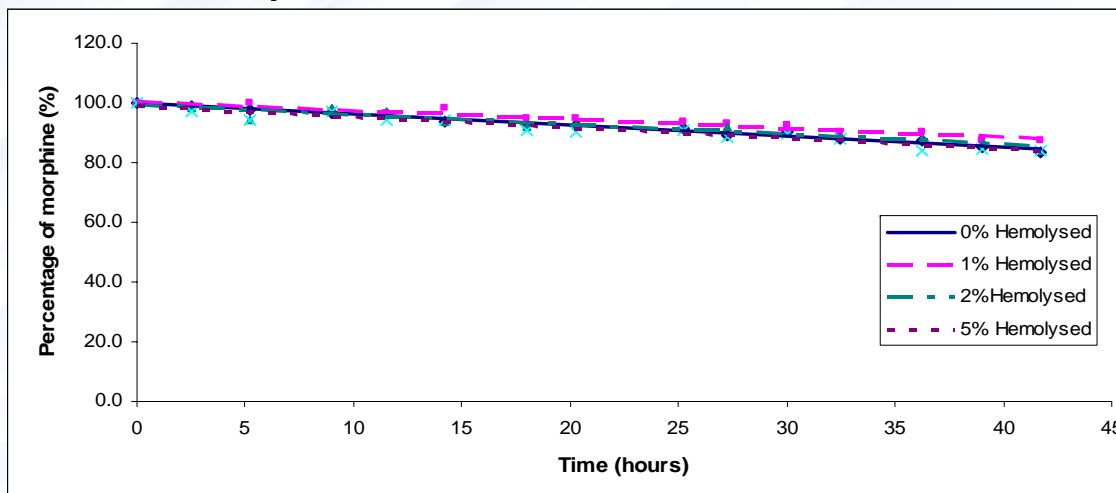
Extracted Blank Reconstituted with a Pure Morphine Solution

A) MeOH/20 mM Ammonium Carbonate Solution



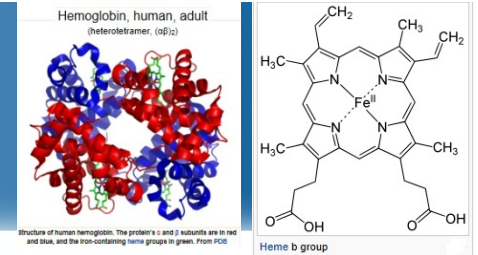
- The morphine peak response drastically decreased in hemolysed samples in the high pH reconstitution solution over time

B) MeOH/Water Solution



- In the MeOH/water mixture, the peak area response was constant regardless of the hemolysis %

Hypothesis

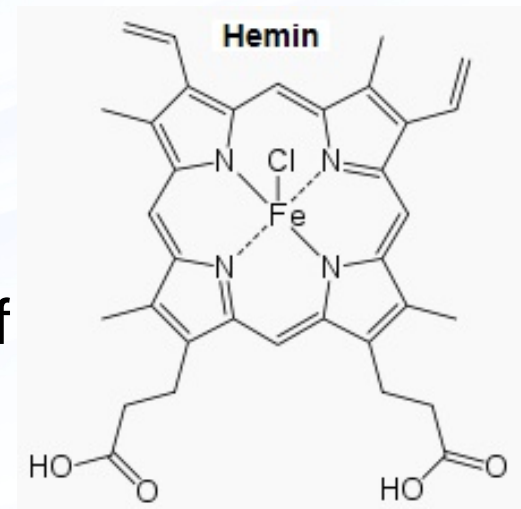


- During plasma generation, hemolysis may occur and produce the presence of free **hemoglobin and heme** in plasma
- **Hemoglobin** can be converted to **methemoglobin**. Normally 2% of hemoglobin is methemoglobin. However, this % can increase in presence of oxidants. Iron in the heme group methemoglobin is in the **Fe³⁺ (ferric) state**, rather than in the Fe²⁺ (ferrous) of normal hemoglobin
- When iron is the ferric (Fe³⁺) state can react **phenols** to form a colored complex.
- Hypothesis: *“Presence of **methemoglobin** in hemolyzed plasma can produce the degradation of compounds containing phenol groups”*

Impact evaluation of Hemoglobin & Methemoglobin presence on PRS

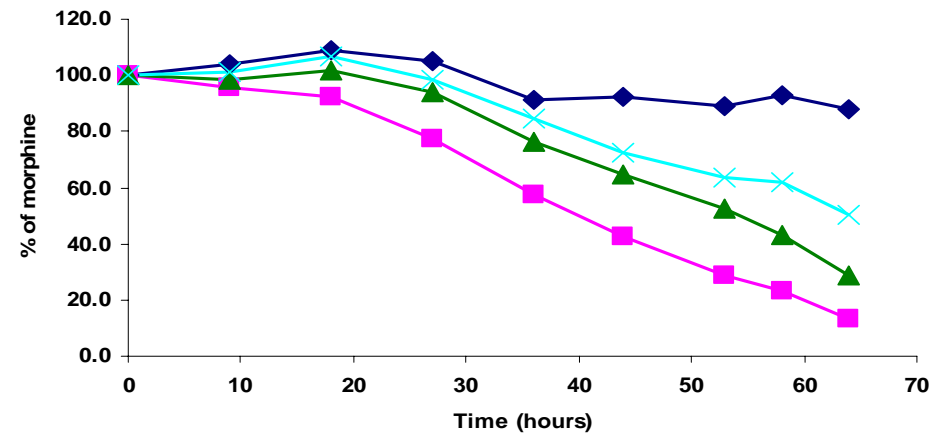
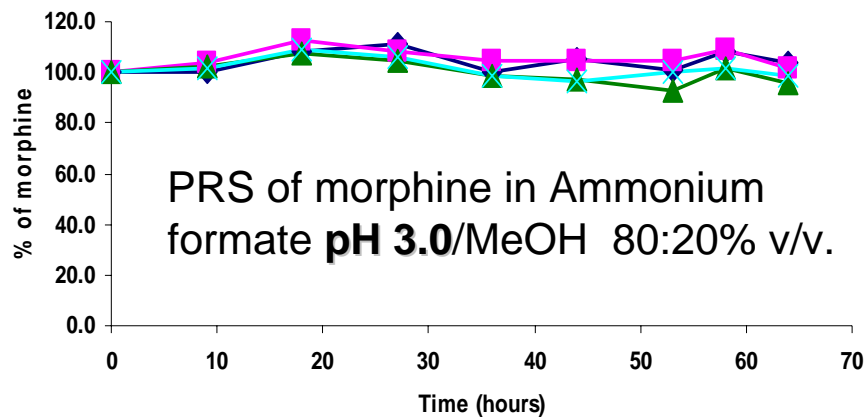
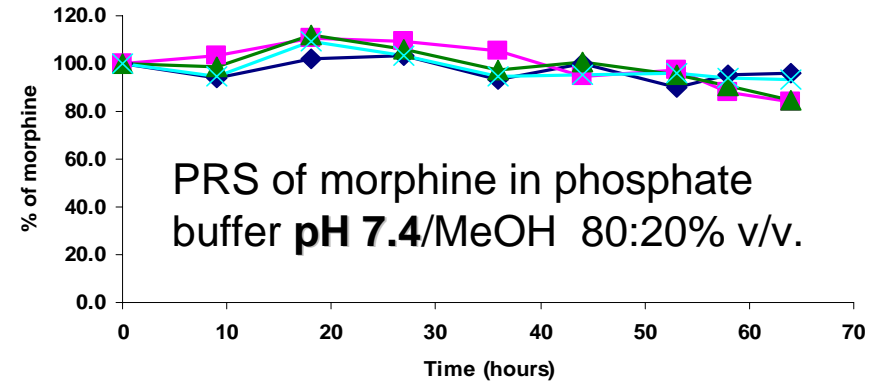
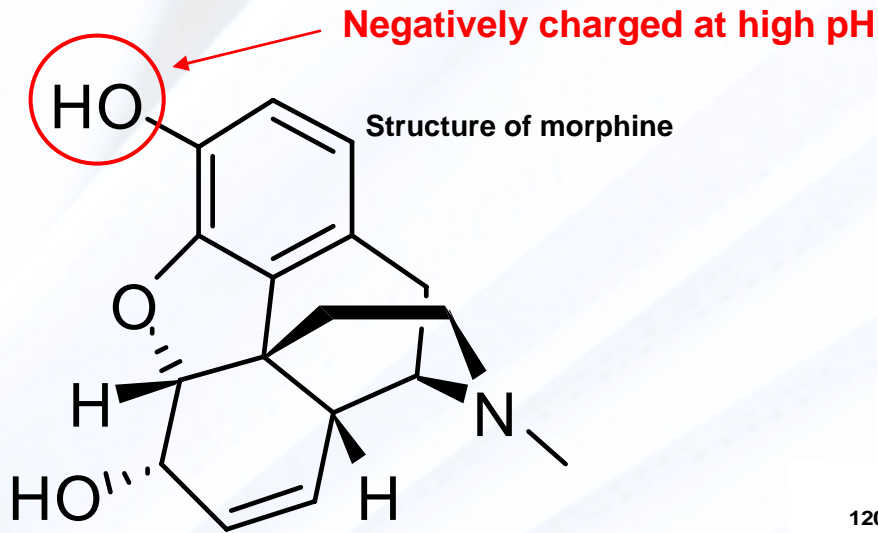
- Blank plasma samples were extracted
 - **7.5%** Hemolysed human plasma
 - 0% Hemolyzed human plasma
 - 0% Hemolyzed human plasma spiked with **1 μ M Hemoglobin**
 - 0% Hemolyzed human plasma spiked with **2 μ M of Hemin**
- Extracted plasma blanks were reconstituted with a pure solution of
 - Morphine, (phenol group)
 - Codeine (methoxy group)
- Test compounds dissolved in a mixture of
 - MeOH/ammonium formate **pH 3** or
 - MeOH/Phosphate buffer **pH 7.4** or
 - MeOH/Ammonium bicarbonate **pH 10**

Heme containing a **ferric iron** ion with a chloride ligand



PRS of morphine in different pH buffer

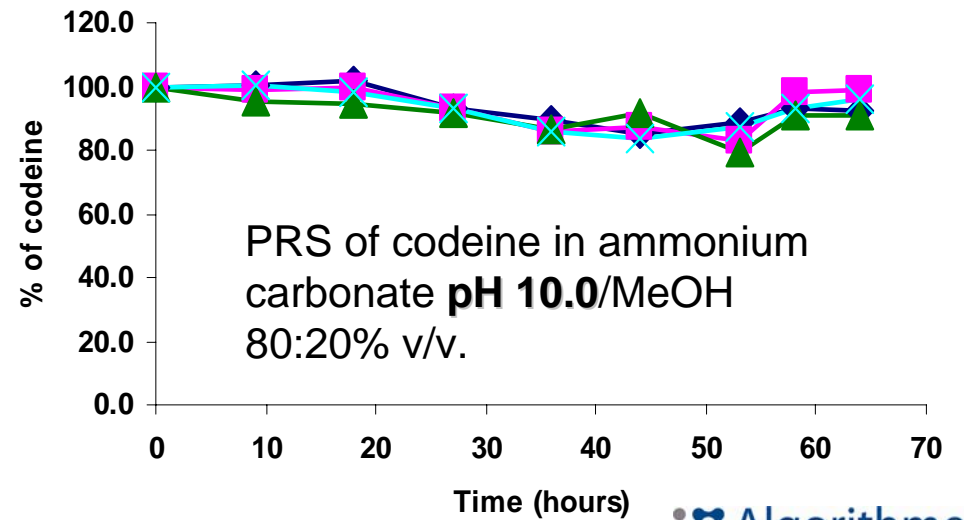
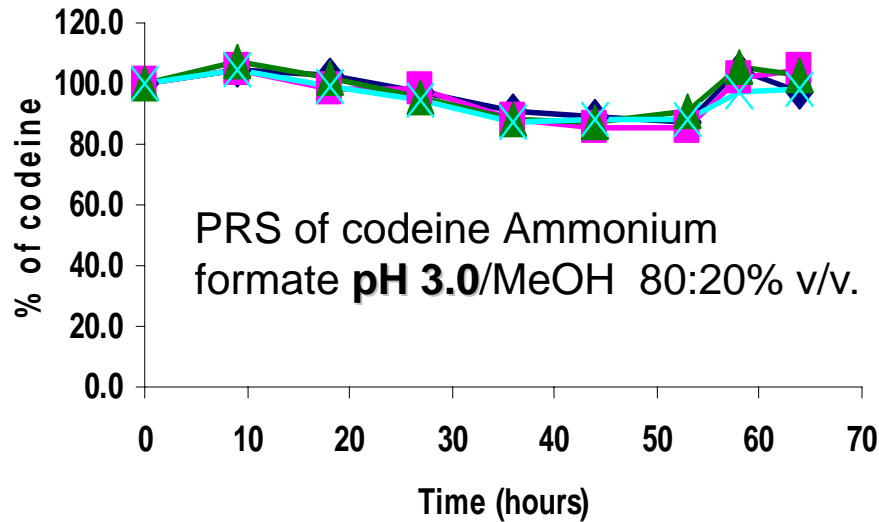
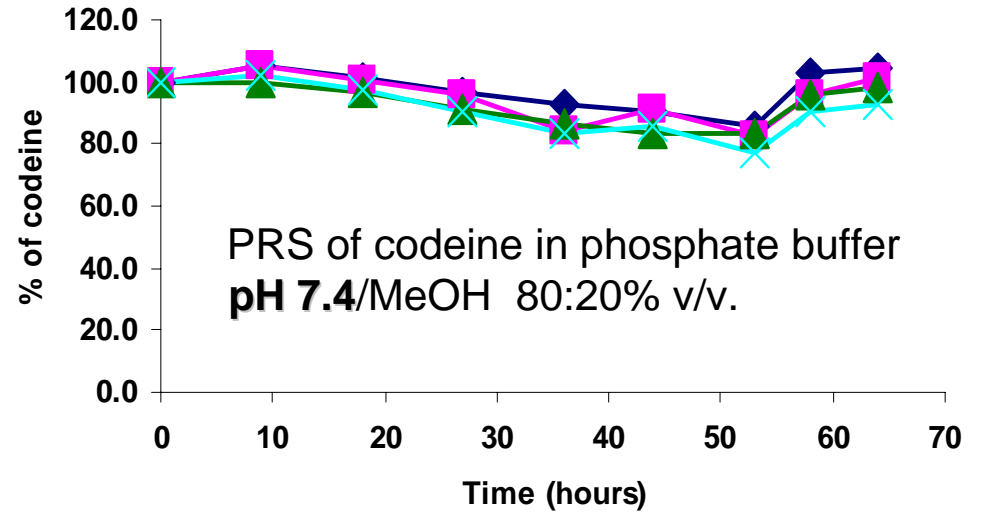
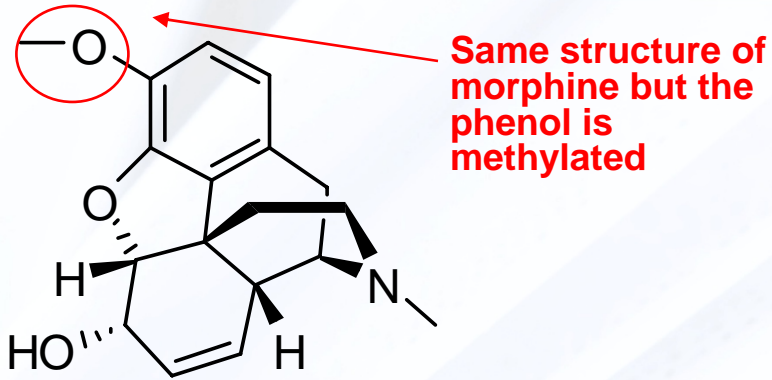
- ◆ Human plasma
- hemolyzed 7.5%
- ▲ plasma+2 μ M Hemin
- ✕ plasma+ 1 μ M hemoglobin



PRS of Codeine in different pH buffer

- ◆ Human plasma
- hemolyzed 7.5%
- ▲ plasma+2 μM Herin
- ✧ plasma+ 1 μM hemoglobin

Structure of codeine



Conclusions

- In the present work, we have clearly demonstrated that sample hemolysis **can impact the stability**.
- Therefore, to ensure the **integrity** of hemolyzed incurred samples, the evaluation of the **stability** of the analyte in this particular matrix **must be investigated**.
- The post-extract stability of phenolic compounds can be pro-actively overcome by having a reconstitution solution at a **pH below the pKa of the functional group**.

“The potential impact of hemolysis on drug stability should be evaluated and resolved at an early stage of the method development to avoid the generation of inaccurate data during incurred sample analysis.”

“Impact of Sample Hemolysis on Drug Stability in Regulated Bioanalysis”

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Bioanalysis, Sep 2011, Vol. 3, No. 18, Pages 2097-2105.